

1. IDENTIFICATION

Product Name Potassium Silicate Solutions (MR 1.6 - 2.6)

Other Names AGSIL 32; Potash Water Glass Solution; Soluble Potash Gass

Uses Agriculture applications.

Chemical Family No Data Available

Chemical FormulaxSiO2/K2O (x ranges from 1.6 - 2.6)Chemical NameSilicic acid, potassium salt, solution

Product Description Liquid potassium silicate.

Contact Details of the Supplier of this Safety Data Sheet

 Organisation
 Location
 Telephone

 Redox Ltd
 2 Swettenham Road
 +61-2-97333000

Minto NSW 2566 Australia

Redox Ltd 11 Mayo Road +64-9-2506222

Wiri Auckland 2104 New Zealand

Redox Inc. 3960 Paramount Boulevard +1-424-675-3200

Suite 107

Lakewood CA 90712

USA

Redox Chemicals Sdn Bhd Level 2, No. 8, Jalan Sapir 33/7 +60-3-5614-2111

Seksyen 33, Shah Alam Premier Industrial Park

40400 Shah Alam Sengalor, Malaysia

Emergency Contact Details

For emergencies only; DO NOT contact these companies for general product advice.

Organisation Location Telephone Poisons Information Centre Westmead NSW 1800-251525 131126 Chemcall Australia 1800-127406 +64-4-9179888 +64-4-9179888 Chemcall Malaysia Chemcall New Zealand 0800-243622 +64-4-9179888 National Poisons Centre New Zealand 0800-764766 CHEMTREC USA & Canada 1-800-424-9300 CN723420

2. HAZARD IDENTIFICATION

Poisons Schedule (Aust) Schedule 5

+1-703-527-3887



Globally Harmonised System

Hazard Classification Hazardous according to the criteria of the Globally Harmonised System of Classification and Labelling of

Chemicals (GHS)

Hazard Categories Acute Toxicity (Oral) - Category 5

Skin Corrosion/Irritation - Category 2 Serious Eye Damage/Irritation - Category 1

Pictograms

Signal Word Danger

Hazard Statements H303 May be harmful if swallowed.

H315 Causes skin irritation.

H318 Causes serious eye damage.

Precautionary Statements Prevention **P280** Wear protective gloves/eye protection/face protection.

Response P305 + P351 + P338 + IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses,

P310 if present and easy to do. Continue rinsing. Immediately call a POISON

CENTRE/doctor.

P302 + P352 IF ON SKIN: Wash with plenty of water.

P332 + P313 If skin irritation occurs: Get medical attention.

P312 Call a POISON CENTER or doctor if you feel unwell.

P362 + P364 Take off contaminated clothing and wash it before reuse.

National Transport Commission (Australia)

Australian Code for the Transport of Dangerous Goods by Road & Rail (ADG Code)

Dangerous Goods Classification NOT Dangerous Goods according to the criteria of the Australian Code for the Transport of Dangerous Goods

by Road & Rail (ADG Code)

3. COMPOSITION/INFORMATION ON INGREDIENTS

Ingredients

Chemical Entity	Formula	CAS Number	Proportion
Potassium silicate	Unspecified	1312-76-1	30 - 60 %
Water	H20	7732-18-5	30 - 60 %

4. FIRST AID MEASURES

Description of necessary measures according to routes of exposure

Swallowed IF SWALLOWED: Rinse mouth thoroughly with water, then give water to drink. Do not induce vomiting. If vomiting occurs,

drink more water to further dilute the product. Call a Poison Centre or doctor/physician for advice.

Eye IF IN EYES: Immediately flush eyes with running water for several minutes, holding eyelids open and occasionally lifting

the upper and lower lids. Remove contact lenses if present and easy to do. Continue rinsing for at least 15 minutes.

Immediately call a Poison Centre or doctor/physician for advice.

Skin IF ON SKIN: Immediately flush skin with running water for at least 20 minutes, while removing contaminated clothing and

> shoes. If skin irritation occurs, get medical advice/attention. Wash contaminated clothing and shoes before reuse. *No attempt should be made to neutralize the alkali with acid solutions, as this could aggravate the burns.

Inhaled IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing until recovered. If

respiratory symptoms persist, get medical advice/attention.

Advice to Doctor Treat symptomatically as for strong alkalis.

Medical Conditions Aggravated by No information available.

Exposure

5. FIRE FIGHTING MEASURES

General Measures If safe to do so, move undamaged containers from fire area. Cool containers with water spray until well after fire is out.

Flammability Conditions Non-combustible under normal conditions of use (aqueous solution).

Extinguishing Media If material is involved in a fire, use dry chemical, Carbon dioxide (CO2), foam or water spray for extinction. No media

identified as unsuitable.

Fire and Explosion Hazard Flammable hydrogen gas may be produced on prolonged contact with metals such as aluminium, tin, lead and zinc.

Hazardous Products of

Fire or heat may produce irritating, toxic and/or corrosive fumes, including Potassium silicate containing mists.

Combustion

Special Fire Fighting Instructions

Contain runoff from fire control or dilution water - Runoff may cause pollution.

Personal Protective Equipment Wear positive pressure self-contained breathing apparatus (SCBA). Structural firefighters' protective clothing will only

provide limited protection.

Flash Point No Data Available No Data Available **Lower Explosion Limit Upper Explosion Limit** No Data Available **Auto Ignition Temperature** No Data Available **Hazchem Code** No Data Available

6. ACCIDENTAL RELEASE MEASURES

General Response Procedure Ensure adequate ventilation. ELIMINATE all ignition sources. Do not touch or walk through spilled material - Spilled liquids

are very slippery! Avoid breathing any fumes formed, and contact with eyes, skin and clothing.

*Dries to form glass film which can easily cut skin.

Clean Up Procedures Spilled liquid may be collected using a vacuum truck. Absorb remaining liquid with earth, sand or other non-combustible

material and transfer to a suitable container for disposal (see SECTION 13).

Containment Stop leak if you can do it without risk. Prevent entry into waterways, sewers, basements or confined areas. Dike far

ahead of large spill for later disposal.

Decontamination If containment is impossible, neutralize contaminated area and flush with large quantities of water.

Environmental Precautionary

Measures

Prevent entry into drains and waterways.

Evacuation Criteria Spill or leak area should be isolated immediately. Keep unauthorised personnel away.

7. HANDLING AND STORAGE

Handling Safety showers and eyewash facilities should be provided within the immediate work area for emergency use. Ensure

adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Avoid generating and inhaling vapours/spray mist, and contact with eyes, skin and clothing. Do not ingest. Use personal protective equipment as required (see SECTION 8). Avoid overheating (decomposition). Take appropriate precautions when handling (bulk) product whilst hot as it can cause thermal burns. Ensure material is used in an appropriately bunded area to prevent

release into soil, drains and waterways.

Storage Store in a cool, dry and well-ventilated place, out of direct sunlight. Ensure containers are labelled and kept closed when

not in use; Promptly clean residue from closures with cloth. Avoid exposure to air. Avoid prolonged storage above 50 °C or below 10 °C. Store away from foodstuffs and incompatible materials (see SECTION 10). Store in accordance with all

local regulations and codes of practice.

Storage temperature: 0 - 70 °CLoading temperature: 10 - 50 °C

ContainerStore in clean steel or plastic containers. Store in clean steel or plastic containers. Mild steel is the most suitable material of construction for drums, tanks, valves, pipe-work, etc. Concrete storage tanks can be used but must be strong enough

to hold the weight of Potassium Silicate solution to be stored and thick enough to prevent seepage of water.

8. EXPOSURE CONTROLS / PERSONAL PROTECTION

General No specific exposure standards are available for this product. Manufacturers recommended limit for good practice (by

analogy with Potassium hydroxide):

- Safe Work Australia Exposure Standard: TWA = 2 mg/m3 Peak limitation.

*Peak limitation means a maximum or peak airborne concentration of a particular substance determined over the shortest

analytically practicable period of time, which does not exceed 15 minutes.

Exposure Limits No Data Available

Biological Limits No information available.

Engineering Measures Ensure exposure is managed within recommended exposure limits. A system of local and/or general exhaust is

recommended to keep employee exposures as low as possible. Local exhaust ventilation is generally preferred because

it can control the emissions of the contaminant at its source, preventing dispersion of it into the general work area.

Personal Protection Equipment - Respiratory protect

- Respiratory protection: Respiratory protection is not normally required due to low inhalation risk. If material is likely to be vaporized the use an approved respirator is necessary. Recommended: Consult a respiratory equipment supplier to

aid selection of the appropriate type.

- Eye/face protection: Wear appropriate eye protection to prevent eye contact. Recommended: Wear glasses with side shields. If contact with material is likely, the use of chemical resistant goggles in combination with a full face shield is

recommended.

- Hand protection: Wear protective gloves. Recommended: Wear chemical resistant gloves. If contact is likely, the use of

full arm length gauntlets is recommended.

- Skin/body protection: Wear appropriate personal protective clothing to avoid skin contact. Recommended: Wear chemical resistant overalls, a full apron, or similar protective clothing, and chemical resistant protective boots.

Special Hazards PrecaustionsThe use of protective clothing and equipment depends on the degree and nature of exposure. Dried silicate can present

physical hazards including cuts and abrasions. Wear cut resistant gloves if handling dried silicate.

Work Hygienic Practices Wash thoroughly after handling. Do not eat, drink or smoke when using this product. Wash contaminated clothing and

protective equipment before storing and re-using. The use of barrier cream is recommended.

9. PHYSICAL AND CHEMICAL PROPERTIES

Physical StateLiquidAppearanceThick liquidOdourOdourless

Colour Clear to hazy, colourless pH 11 - 13 (of the concentrate)

Vapour Pressure No Data Available
Relative Vapour Density No Data Available

Boiling Point 105 - 108 °C **Melting Point** approx. 0 °C **Freezing Point** No Data Available Solubility Soluble in water

Specific Gravity 1.2 - 1.6 (typical range) **Flash Point** No Data Available No Data Available **Auto Ignition Temp Evaporation Rate** No Data Available No Data Available **Bulk Density Corrosion Rate** No Data Available **Decomposition Temperature** 102 - 108 °C

No Data Available Density **Specific Heat** No Data Available **Molecular Weight** No Data Available **Net Propellant Weight** No Data Available **Octanol Water Coefficient** No Data Available **Particle Size** No Data Available **Partition Coefficient** No Data Available **Saturated Vapour Concentration** No Data Available

Viscosity 20 - 5,000 cps (@ No Data Available)

Volatile Percent 30 - 60% **VOC Volume** 0 %

Additional Characteristics Absorbs Carbon dioxide on exposure to air, which results in the deposition of insoluble silica.

Potential for Dust Explosion Not applicable.

Fast or Intensely Burning

Vapour Temperature

Characteristics

No information available.

No Data Available

Flame Propagation or Burning

Rate of Solid Materials

No information available.

Non-Flammables That Could

Contribute Unusual Hazards to a

Fire

No information available.

Properties That May Initiate or Contribute to Fire Intensity

Non-combustible under normal conditions of use (aqueous solution).

Reactions That Release Gases or

Vapours

Fire or heat may produce irritating, toxic and/or corrosive fumes, including Potassium silicate containing mists.

Release of Invisible Flammable

Vapours and Gases

Flammable hydrogen gas will form on reaction with aluminium, copper, zinc, etc.

10. STABILITY AND REACTIVITY

General Information Strongly alkaline; Will react exothermically with acids. Gels and generates heat when mixed with acid. May react with

ammonium salts resulting in evolution of ammonia gas. Can etch glass if not promptly removed.

Chemical Stability Stable in sealed containers.

Conditions to Avoid Avoid exposure to air. Avoid prolonged storage above 50 °C or below 10 °C.

Materials to Avoid Incompatible/reactive with acids, aluminium, copper, brass, bronze, zinc, tin and lead.

Hazardous Decomposition

Products

The solution will boil if overheated, and irritating Potassium silicate containing mists will be released. Flammable

hydrogen gas will form on reaction with aluminium, copper, zinc, etc.

Hazardous Polymerisation

No information available.

11. TOXICOLOGICAL INFORMATION

General Information

- Acute toxicity: May be harmful if swallowed. Swallowing can result in nausea, vomiting, abdominal pain and diarrhoea; May cause severe irritation to the mouth, throat and stomach.
- Skin corrosion/irritation: Causes skin irritation. May cause itching and skin rash. Prolonged or repeated skin contact may cause dry skin. Defatting of the skin can result in irritation and dermatitis (inflammation of the skin).
- Eye damage/irritation: Causes serious eye damage. A severe eye irritant; May cause conjunctivitis (inflammation of the eyes) and possibly corneal burns and ulceration.
- Respiratory/skin sensitisation: Not considered to be a skin sensitiser. Not sensitising (Mouse LLNA Sodium metasilicate).
- Germ cell mutagenicity: There is no evidence of a genotoxic potential for soluble silicates.
- Carcinogenicity: The information available does not indicate any potential for carcinogenicity.
- Reproductive toxicity: No indications of reproductive effects for silicates have been reported.
- STOT (single exposure): Exposure to vapours at room temperature is an unlikely route of exposure due to its low vapour pressure. Spray mist will cause respiratory irritation and may result in coughing as well as inflammation of nose, throat and windpipe.
- STOT (repeated exposure): Frequent ingestion over extended periods of time of gram quantities of silicates is associated with the formation kidney stones and other siliceous urinary calculi in humans.
- Aspiration toxicity: No information available.

Acute

Ingestion Acute toxicity (Oral):

- LD50, Rat: >3,000 mg/kg [By analogy: 30-60% Sodium silicate].

Carcinogen Category

None

12. ECOLOGICAL INFORMATION

Ecotoxicity Acute toxicity testing in fish, invertebrates and algae indicate a low order of toxicity: the soluble silicates exhibit aquatic

toxicities in excess of 100 mg/l irrespective of molar ratio or metal cation.

Persistence/Degradability This material is not persistent in aquatic systems; Diluted material rapidly depolymerizes to yield dissolved silica in a form

that is indistinguishable from natural dissolved silica. It does not contribute to BOD.

Mobility Soluble in water (Sinks and mixes with water). Expected to be mobile in soil.

Environmental Fate This material is not acutely toxic in aquatic systems, but its high pH when undiluted or un-neutralized is acutely harmful to

aquatic life. Avoid contaminating waterways.

Bioaccumulation Potential

This material does not bioaccumulate except in species that use silica as a structural material such as diatoms and

siliceous sponges. Neither silica nor sodium will appreciably bioconcentrate up the food chain.

Environmental Impact No Data Available

13. DISPOSAL CONSIDERATIONS

General Information Dispose of contents/container in accordance with local/regional/national regulations.

Special Precautions for Land Fill Normally suitable for disposal at approved land waste site after dilution or neutralisation. Not suitable for incineration.

14. TRANSPORT INFORMATION

Land Transport (Australia)

ADG Code

Proper Shipping Name Potassium Silicate Solutions (MR 1.6 - 2.6)

Class No Data Available
Subsidiary Risk(s) No Data Available

No Data Available

UN Number No Data Available
Hazchem No Data Available
Pack Group No Data Available
Special Provision No Data Available

Comments NON-DANGEROUS GOODS: Not regulated for LAND transport.

Land Transport (Malaysia)

ADR Code

Proper Shipping Name Potassium Silicate Solutions (MR 1.6 - 2.6)

Class No Data Available
Subsidiary Risk(s) No Data Available

No Data Available

UN NumberNo Data AvailableHazchemNo Data AvailablePack GroupNo Data AvailableSpecial ProvisionNo Data Available

Comments NON-DANGEROUS GOODS: Not regulated for LAND transport.

Land Transport (New Zealand)

NZS5433

Proper Shipping Name Potassium Silicate Solutions (MR 1.6 - 2.6)

Class No Data Available
Subsidiary Risk(s) No Data Available
No Data Available

UN Number No Data Available
Hazchem No Data Available
Pack Group No Data Available
Special Provision No Data Available

Comments NON-DANGEROUS GOODS: Not regulated for LAND transport.

Land Transport (United States of America)

US DOT

UN Number

Hazchem

Proper Shipping Name Potassium Silicate Solutions (MR 1.6 - 2.6)

Class No Data Available
Subsidiary Risk(s) No Data Available
No Data Available

No Data Available No Data Available

Pack GroupNo Data AvailableSpecial ProvisionNo Data Available

Comments NON-DANGEROUS GOODS: Not regulated for LAND transport.

Sea Transport

IMDG Code

Proper Shipping Name Potassium Silicate Solutions (MR 1.6 - 2.6)

Class No Data Available
Subsidiary Risk(s) No Data Available
UN Number No Data Available
Hazchem No Data Available
Pack Group No Data Available
Special Provision No Data Available
EMS No Data Available

Marine Pollutant No

Comments NON-DANGEROUS GOODS: Not regulated for SEA transport.

Air Transport

IATA DGR

Proper Shipping Name Potassium Silicate Solutions (MR 1.6 - 2.6)

Class No Data Available
Subsidiary Risk(s) No Data Available
UN Number No Data Available
Hazchem No Data Available
Pack Group No Data Available
Special Provision No Data Available

Comments NON-DANGEROUS GOODS: Not regulated for AIR transport.

National Transport Commission (Australia)

Australian Code for the Transport of Dangerous Goods by Road & Rail (ADG Code)

Dangerous Goods Classification NOT Dangerous Goods according to the criteria of the Australian Code for the Transport of Dangerous Goods

by Road & Rail (ADG Code)

15. REGULATORY INFORMATION

General InformationNo Data AvailablePoisons Schedule (Aust)Schedule 5

Environmental Protection Authority (New Zealand)

Hazardous Substances and New Organisms Amendment Act 2015

Approval Code Additives Process Chemicals and Raw Materials Subsidiary Hazard Group Standard 2020 HSR002503

*HSR004658 (Revoked)

National/Regional Inventories

Australia (AIIC) Listed

Canada (DSL) Not Determined

Canada (NDSL) Not Determined

China (IECSC) Not Determined

Europe (EINECS) Not Determined

Europe (REACh) Not Determined

Japan (ENCS/METI) Not Determined

Korea (KECI) Not Determined

Malaysia (EHS Register) Not Determined

New Zealand (NZIoC) Listed

Philippines (PICCS) Not Determined

Switzerland (Giftliste 1) Not Determined

Switzerland (Inventory of Notified

Substances)

Not Determined

Taiwan (NCSR) Not Determined

USA (TSCA) Not Determined

16. OTHER INFORMATION

Related Product Codes POTSIL5000, POTSIL5000, POTSIL6000, POTSIL6050, POTSIL6051, POTSIL7000

Revision

AICS Australian Inventory of Chemical Substances

atm Atmosphere

CAS Chemical Abstracts Service (Registry Number)

cm² Square CentimetresCO2 Carbon Dioxide

COD Chemical Oxygen Demand **deg C (°C)** Degrees Celcius

EPA (New Zealand) Environmental Protection Authority of New Zealand

deg F (°F) Degrees Farenheit

g Grams

g/cm³ Grams per Cubic Centimetre

g/I Grams per Litre

HSNO Hazardous Substance and New Organism **IDLH** Immediately Dangerous to Life and Health **immiscible** Liquids are insoluable in each other.

inHg Inch of Mercury inH2O Inch of Water

K Kelvin **kg** Kilogram

kg/m³ Kilograms per Cubic Metre

Ib Pound

LC50 LC stands for lethal concentration. LC50 is the concentration of a material in air which causes the death of 50%

(one half) of a group of test animals. The material is inhaled over a set period of time, usually 1 or 4 hours.

LD50 LD stands for Lethal Dose. LD50 is the amount of a material, given all at once, which causes the death of 50% (one half) of a group of test animals.

Itr or L Litre

m³ Cubic Metre

mbar Millibar

mg Milligram

mg/24H Milligrams per 24 Hours

mg/kg Milligrams per Kilogram

mg/m³ Milligrams per Cubic Metre

Misc or Miscible Liquids form one homogeneous liquid phase regardless of the amount of either component present.

mm Millimetre

mmH20 Millimetres of Water

mPa.s Millipascals per Second

N/A Not Applicable

NIOSH National Institute for Occupational Safety and Health

NOHSC National Occupational Heath and Safety Commission

OECD Organisation for Economic Co-operation and Development

Oz Ounce

PEL Permissible Exposure Limit

Pa Pascal

ppb Parts per Billion

ppm Parts per Million

ppm/2h Parts per Million per 2 Hours

ppm/6h Parts per Million per 6 Hours

psi Pounds per Square Inch

R Rankine

RCP Reciprocal Calculation Procedure

STEL Short Term Exposure Limit

TLV Threshold Limit Value

tne Tonne

TWA Time Weighted Average

ug/24H Micrograms per 24 Hours

UN United Nations

wt Weight