

1. IDENTIFICATION

Product Name	Hydrofluoric Acid (>=25 - <=50%)
Other Names	Hydrogen fluoride, aqueous solution
Uses	Chemical intermediate, etching and polishing of glass and pottery, cleaning of metals, mineral extraction.
Chemical Family	No Data Available
Chemical Formula	HF.H2O
Chemical Name	Hydrofluoric acid, aqueous solution
Product Description	No Data Available

Contact Details of the Supplier of this Safety Data Sheet

Organisation	Location	Telephone
Redox Ltd	2 Swettenham Road Minto NSW 2566 Australia	+61-2-97333000
Redox Ltd	11 Mayo Road Wiri Auckland 2104 New Zealand	+64-9-2506222
Redox Inc.	3960 Paramount Boulevard Suite 107 Lakewood CA 90712 USA	+1-424-675-3200
Redox Chemicals Sdn Bhd	Level 2, No. 8, Jalan Sapir 33/7 Seksyen 33, Shah Alam Premier Industrial Park 40400 Shah Alam Sengalor, Malaysia	+60-3-5614-2111

Emergency Contact Details

For emergencies only; DO NOT contact these companies for general product advice.

Organisation	Location	Telephone
Poisons Information Centre	Westmead NSW	1800-251525 131126
Chemcall	Australia	1800-127406 +64-4-9179888
Chemcall	Malaysia	+64-4-9179888
Chemcall	New Zealand	0800-243622 +64-4-9179888
National Poisons Centre	New Zealand	0800-764766
CHEMTREC	USA & Canada	1-800-424-9300 CN723420 +1-703-527-3887

2. HAZARD IDENTIFICATION

Poisons Schedule (Aust)

Schedule 7

Redox Ltd Corporate Office Sydney Locked Bag 15 Minto NSW 2566 Australia 2 Swettenham Road Minto NSW 2566 Australia

All Deliveries: 4 Holmes Road Minto NSW 2566 Australia

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 Phone
 +61 2 9733 3000

 Fax
 +61 2 9733 3111

 E-mail
 sydney@redox.com

 Web
 www.redox.com

 ABN
 92 000 762 345

AustraliaNew ZealandAdelaideAucklandBrisbaneChristchurchMelbourneHawke's BayPerthUKSydneyLondon

 New Zealand
 Malaysia

 Auckland
 Kuala Lumpur

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Globally Harmonised System

Hazard Classification		Hazardous according to Chemicals (GHS)	the criteria of the Globally Harmonised System of Classification and Labelling of	
Hazard Categories Con		Corrosive to Metals - Ca	Corrosive to Metals - Category 1	
		Acute Toxicity (Oral) - Category 2		
		Acute Toxicity (Dermal)	- Category 1	
		Acute Toxicity (Inhalation) - Category 2		
		Skin Corrosion/Irritation - Category 1A		
		Serious Eye Damage/Irr	itation - Category 1	
		Specific Target Organ Toxicity (Single Exposure) - Category 1		
		Specific Target Organ T	oxicity (Repeated Exposure) - Category 1	
Pictograms				
Signal Word		Danger		
Hazard Statements		H290	May be corrosive to metals.	
		H300 + H310 + H330	Fatal if swallowed, in contact with skin or if inhaled.	
		H314	Causes severe skin burns and eye damage.	
		H370	Causes damage to organs.	
		H372	Causes damage to organs through prolonged or repeated exposure.	
Precautionary Statements	Prevention	P262	Do not get in eyes, on skin, or on clothing.	
		P280	Wear protective gloves/protective clothing/eye protection/face protection.	
		P260	Do not breathe mist/vapours.	
		P284	Wear respiratory protection.	
		P270	Do not eat, drink or smoke when using this product.	
		P271	Use only outdoors or in a well-ventilated area.	
		P234	Keep only in original packaging.	
	Response	P310	Immediately call a POISON CENTER or doctor.	
		P304 + P340	IF INHALED: Remove victim to fresh air and keep comfortable for breathing.	
		P303 + P361 + P353	IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water or shower.	
		P305 + P351 + P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.	
		P363	Wash contaminated clothing before reuse.	
		P301 + P330 + P331	IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.	
		P390	Absorb spillage to prevent material-damage.	
		P321	Specific treatment (see First Aid Measures on Safety Data Sheet).	
	Storage	P403 + P233	Store in a well-ventilated place. Keep container tightly closed.	
		P405	Store locked up.	
	Disposal	P501	Dispose of contents/container in accordance with local / regional / national / international regulations.	

National Transport Commission (Australia)

Australian Code for the Transport of Dangerous Goods by Road & Rail (ADG Code)

Dangerous Goods Classification

Dangerous Goods according to the criteria of the Australian Code for the Transport of Dangerous Goods by Road & Rail (ADG Code)

Safe Work Australia

National Guide for Classifying Hazardous Chemicals under the Model WHS Regulations

Hazard Classification

Hazardous according to the criteria of Safe Work Australia under Model WHS Regulations

3. COMPOSITION/INFORMATION ON INGREDIENTS

Ingredients

Chemical Entity	Formula	CAS Number	Proportion
Hydrofluoric acid	HF	7664-39-3	>=25 - <=50 %
Water	H20	7732-18-5	Balance %

4. FIRST AID MEASURES

Description of necessary measures according to routes of exposure

Swallowed	IF SWALLOWED: Rinse mouth with water, then drink plenty of water. Immediately call a Poison Centre or doctor/physician for emergency medical advice! Do NOT induce vomiting. If vomiting occurs naturally, have victim lean forward to reduce risk of aspiration. Rinse mouth of victim thoroughly with water and spit out rinse water. Never give anything by mouth if victim is losing consciousness, is unconscious or convulsing. Transport to hospital by ambulance. *If victim is conscious, give four 600 mg or five 500 mg effervescent calcium gluconate tablets. Where calcium gluconate tablets are only available in other active strength levels, the total active concentration should be approximately 2,400 - 2,500 mg.
Eye	IF IN EYES: Do not rub affected area! Immediately call a Poison Centre or doctor/physician for emergency medical advice! Immediately flush eyes with running water or copious isotonic saline for several minutes, holding the eyelids open and occasionally lifting the upper and lower lids. Take care not to rinse contaminated water into a non-affected eye. Carefully remove contact lenses if present and easy to do. Continue flushing until advised to stop by a Poisons Information Centre (e.g. phone Australia 13 11 26; New Zealand 0800 764 766) or a doctor, or for at least 15 minutes. Transport to hospital by ambulance; Continue eye irrigation during transport to hospital. An ophthalmologist should always be consulted, as severe corneal damage is possible. Contact with eyes, even for short periods, can cause blindness.
Skin	IF ON SKIN: Immediately call a Poison Centre or doctor/physician for emergency medical advice! Remove and isolate contaminated clothing and shoes. Gently flush contaminated skin with running water for 1- 2 minutes. In case of gross contamination, drench contaminated clothing and skin with plenty of water before removing clothes and footwear. For minor skin contact, avoid spreading material on unaffected skin. Isolate contaminated clothing by sealing in a bag or other container. Wearing clean protective gloves, gently massage Calcium gluconate gel into and around the affected area. If gel is not readily available, continue flushing with water. Transport to hospital by ambulance; Continue application of gel during transport to hospital and for at least 15 minutes after the pain has subsided (this treatment may be required for several hours). Wear clean gloves when applying gel. *For large or severe burns, give four 600 mg or five 500 mg effervescent calcium gluconate tablets by mouth. Where calcium gluconate tablets are only available in other active strength levels, the total active concentration should be approximately 2,400 - 2,500 mg.
Inhaled	IF INHALED: Immediately call a Poison Centre or doctor/physician for emergency medical advice! Remove victim to fresh air and keep at rest in a position comfortable for breathing. Give artificial respiration if victim is not breathing. Do not use mouth-to-mouth method if victim ingested or inhaled the substance; give artificial respiration with the aid of a pocket mask equipped with a one-way valve or other proper respiratory medical device. Administer oxygen if breathing is difficult. Transport to hospital by ambulance; Continue observation for at least 48 hours due to the danger of pulmonary edema. *If victim is conscious, give four 600 mg or five 500 mg effervescent calcium gluconate tablets. Where calcium gluconate tablets are only available in other active strength levels, the total active concentration should be approximately 2,400 -

2,500 mg.

Ensure that medical personnel are aware of the material(s) involved and take precautions to protect themselves. For advice, contact a Poisons Information Centre (e.g. phone Australia 13 11 26; New Zealand 0800 764 766) or a doctor. This SDS should accompany the affected person to hospital. Hydrofluoric acid penetrates rapidly and deeply below fat layers binding and depleting tissue calcium. Failure to commence the correct medical treatment promptly may be fatal. Intensive care unit facilities are likely to be needed. Keep victim calm and warm. Effects of exposure (inhalation, ingestion or skin contact) to substance may be delayed.

*Most important symptoms and effects, both acute and delayed: Delayed pulmonary oedema is likely with burns to the face or neck. A skin burn involving more than 1% of body area with 50% or more concentration of Hydrofluoric acid, or more than 5% of body area with any lesser concentrations, may be associated with systemic effects. Skin burns may become necrotic and gangrenous and damaged area may spread. Treatment with intravenous Calcium gluconate should commence immediately. Inhalation may lead to chemical pneumonitis, haemorrhagic pulmonary oedema or laryngeal oedema and may be fatal. Be prepared to intubate or perform tracheotomy.

Medical Conditions Aggravated by No information available. Exposure

5. FIRE FIGHTING MEASURES

Advice to Doctor

General Measures	If safe to do so, move undamaged containers from fire area. Cool containers with flooding quantities of water until well after fire is out. Avoid getting water inside containers. Dike fire-control water for later disposal; do not scatter the material. Fight fire from maximum distance or use unmanned hose holders or monitor nozzles. Withdraw immediately in case of rising sound from venting safety devices or discoloration of tank. ALWAYS stay away from tanks engulfed in fire.
Flammability Conditions	Non-combustible; substance itself does not burn but may decompose upon heating to produce corrosive and/or toxic fumes.
Extinguishing Media	Use dry chemical, Carbon dioxide (CO2), dry sand or alcohol-resistant foam or water spray (fog) for extinction - Do not use water jets. *Large fire: Flood fire area with large quantities of water while knocking down vapours with water fog - If insufficient water supply, knock down vapours only.
Fire and Explosion Hazard	Contact with metals may evolve flammable hydrogen gas. Reaction with water may generate heat which will increase the concentration of fumes in the air and present risk of splashing. Containers may explode when heated or contaminated with water. Vapours may accumulate in confined areas. Inhalation, ingestion or contact with substance may cause severe injury or death.
Hazardous Products of Combustion	Fire will produce irritating, corrosive and/or toxic gases, including Hydrogen fluoride.
Special Fire Fighting Instructions	Contain runoff from fire control or dilution water - Runoff may be corrosive and/or toxic and cause pollution.
Personal Protective Equipment	Wear positive pressure self-contained breathing apparatus (SCBA). Wear chemical protective clothing - It may provide little or no thermal protection. Structural firefighters' protective clothing provides limited protection in fire situations ONLY; it is not effective in spill situations where direct contact with the substance is possible.
Flash Point	No Data Available
Lower Explosion Limit	No Data Available
Upper Explosion Limit	No Data Available
Auto Ignition Temperature	No Data Available
Hazchem Code	2X

6. ACCIDENTAL RELEASE MEASURES

General Response Procedure	Ensure adequate ventilation - Ventilate enclosed spaces before entering. ELIMINATE all ignition sources (no smoking, flares, sparks or flames in immediate area). All equipment used when handling the product must be grounded. Do not touch or walk through spilled material. Do not breathe vapours and prevent contact with eyes, skin and clothing - Inhalation, ingestion or contact with substance may cause severe injury or death!
Clean Up Procedures	Take up with liquid-absorbent and neutralising material and transfer to a suitable container for disposal (see SECTION 13).
Containment	Stop leak if safe to do so - Prevent entry into waterways, drains or confined areas.

	*Vapour-suppressing foam may be used to control vapours. Water spray may be used to knock down or divert vapour clouds.
Decontamination	Neutralise HF with calcium hydroxide or lime or HF absorbent (e.g. Chemizorb HF). Contaminated clothing should be washed with bicarbonate of soda solution. Contaminated equipment or surfaces can be neutralised with calcium hydroxide or slaked lime, before being washed with water.
Environmental Precautionary Measures	Spillages and decontamination runoff should be prevented from entering drains and watercourses.
Evacuation Criteria	Spill or leak area should be isolated immediately. Evacuate personnel to safe areas. Keep unauthorised personnel away. Keep upwind and to higher ground. *Large spill: Immediately contact Police or Fire Brigade; Consider downwind evacuation of areas.
Personal Precautionary Measures	Do not touch damaged containers or spilled material unless wearing appropriate protective clothing (see SECTION 8). *Fully encapsulating, vapor protective clothing should be worn for spills and leaks with no fire.

7. HANDLING AND STORAGE

Handling	Safety showers and eyewash facilities should be provided within the immediate work area for emergency use. A supply of Calcium gluconate gel should be kept in an accessible and convenient location. Ensure adequate ventilation - Handle product only in closed system or provide appropriate exhaust ventilation. VERY TOXIC & CORROSIVE: Handle with extreme care and in accordance with good industrial hygiene and safety practice. Avoid generation of mist/vapours/aerosols. Do not breathe mist/vapours/aerosols. Do not get in eyes, on skin or on clothing. Do not ingest. Wear protective gloves/protective clothing/eye protection/face protection and suitable respirator (see SECTION 8). When diluting, always add the product to water. Never add water to the product. Keep away from heat and sources of ignition - No smoking. Use explosion-proof electrical/ventilating/lighting equipment.
Storage	Store in a cool, dry and well-ventilated place, out of direct sunlight. Keep container tightly closed. Keep away from heat and all sources of ignition - No smoking. Keep away from foodstuffs and incompatible materials (see SECTION 10). Store locked up. Store in accordance with local regulations.
Container	Store HF in properly labelled original or plastic (polyethylene, polypropylene or PVC) containers only. Do not store in metal (steel, copper, aluminium) or glass containers.

8. EXPOSURE CONTROLS / PERSONAL PROTECTION

General	COMPONENT: Hydrogen fluoride (CAS No. 7664-39-3): - Safe Work Australia Exposure Standard: TWA = 3 ppm (2.6 mg/m3) Peak limitation (as F). - New Zealand Workplace Exposure Standard [Next review 2023]: Ceiling = 3 ppm (2.6 mg/m3), as F. - NIOSH REL: TWA = 3 ppm (2.5 mg/m3); 6 ppm (5 mg/m3) 15-minute Ceiling. - OSHA PEL: TWA = 3 ppm. *Immediately dangerous to life or health (IDLH) concentration: 30 ppm.
Exposure Limits	No Data Available
Biological Limits	No information available.
Engineering Measures	All HF work involving the release of HF should be conducted in a scrubbed fume cupboard. The occupational exposure limit value should not be exceeded during any part of the working exposure.
Personal Protection Equipment	 Respiratory protection: Wear respiratory protection. Recommended: Filter type E-P3 or HF or supplied-air respirator when mist/vapours/aerosols are generated (refer to AS/NZS 1715 & 1716). Eye/face protection: Wear appropriate eye protection to prevent eye contact. Recommended: Tightly fitting safety goggles; Face shield. Hand protection: Wear protective gloves. Recommended: Elbow-length impervious gloves. Skin/body protection: Wear appropriate personal protective clothing to prevent skin contact. Recommended: Acid-resistant protective clothing; Chemical-resistant apron; rubber boots.
Special Hazards Precaustions	HF should only be handled by workers who have been adequately trained and assessed as competent in its use. Do not work alone with HF. Specific treatment is necessary in case of poisoning with this substance; the appropriate means with instructions must be available. Ensure that a first aid kit has the necessary content, is kept up to date and is complete to adequately respond to potential HF exposure. Obtain a supply of calcium gluconate gel.

Work Hygienic Practices

Do not eat, drink or smoke when using this product. Do not get in eyes, on skin or on clothing. Wash hands and face thoroughly after handling. Remove/take off immediately all contaminated clothing. Isolate contaminated clothing by sealing in a bag or other container. Decontaminate and wash contaminated clothing and protective equipment before storage or reuse.

9. PHYSICAL AND CHEMICAL PROPERTIES

Physical State	Liquid
Appearance	Liquid
Odour	Sharp, pungent
Colour	Colourless
рН	<1
Vapour Pressure	No Data Available
Relative Vapour Density	No Data Available
Boiling Point	20 ℃
Melting Point	No Data Available
Freezing Point	-83 °C
Solubility	Miscible with water
Specific Gravity	1.1 - 1.22 (Water = 1)
Flash Point	No Data Available
Auto Ignition Temp	No Data Available
Evaporation Rate	No Data Available
Bulk Density	No Data Available
Corrosion Rate	No Data Available
Decomposition Temperature	No Data Available
Density	No Data Available
Specific Heat	No Data Available
Molecular Weight	No Data Available
Net Propellant Weight	No Data Available
Octanol Water Coefficient	No Data Available
Particle Size	No Data Available
Partition Coefficient	No Data Available
Saturated Vapour Concentration	No Data Available
Vapour Temperature	No Data Available
Viscosity	No Data Available
Volatile Percent	No Data Available
VOC Volume	No Data Available
Additional Characteristics	No information available.
Potential for Dust Explosion	Not applicable.
Fast or Intensely Burning Characteristics	No information available.
Flame Propagation or Burning Rate of Solid Materials	No information available.
Non-Flammables That Could Contribute Unusual Hazards to a Fire	Reaction with water may generate heat which will increase the concentration of fumes in the air and present risk of splashing.
Properties That May Initiate or Contribute to Fire Intensity	Non-combustible; Does not burn; However, many reactions may cause fire or explosion.

Reactions That Release Gases or Reacts violently with many compounds producing toxic and corrosive gases, including Hydrogen fluoride. Vapours

Release of Invisible Flammable Reacts violently with bases and is corrosive to most common metals forming flammable/explosive Hydrogen gas. Vapours and Gases

10. STABILITY AND REACTIVITY

General Information	The substance is a weak acid. Reacts violently with many compounds, generating fire and explosion hazard. It reacts violently with bases and is corrosive to most common metals. Attacks glass, some forms of plastic, rubber and coatings.
Chemical Stability	Material is stable under normal conditions of temperature and pressure.
Conditions to Avoid	Avoid generation of mist/vapours/aerosols. Keep away from heat and all sources of ignition.
Materials to Avoid	Incompatible/reactive with alkali metals, fluorine, organic substances, vinyl acetate; potassium permanganate, alkali hydroxides, strong alkalis, fluorides, potassium, metals, sodium, methanesulfonic acid; nitric acid, with, glycerol; acetic anhydride, ammonia, ammonium hydroxide, sodium hydroxide, fuming sulfuric acid, oxides of phosphorus, silicon compounds, sulphuric acid, bismuth acid, ethanolamine.
Hazardous Decomposition Products	Reacts violently with many compounds. producing toxic and corrosive gases, including Hydrogen gas, Hydrogen fluoride.
Hazardous Polymerisation	No information available.

11. TOXICOLOGICAL INFORMATION

General Information	Information on toxicological effects: - Acute toxicity: Fatal if swallowed, in contact with skin and if inhaled. - Skin corrosion/irritation: Causes severe skin burns and eye damage. - Eye damage/irritation: Causes serious eye damage. - Respiratory/skin sensitisation: No information available. - Germ cell mutagenicity: No information available. - Carcinogenicity: Fluorides (inorganic) are Classified by the IARC Monographs as "Not classifiable as to its carcinogenicity to humans" (Group 3). - Reproductive toxicity: No information available. - STOT (single exposure): Causes damage to organs (respiratory system). Corrosive to the respiratory tract. - STOT (single exposure): Causes damage to organs through prolonged or repeated exposure (nervous system; respiratory system; teeth; bone). In humans, skeletal fluorosis is a known effect of prolonged intake of fluoride, by both oral and inhalation routes. - Aspiration toxicity: No information available. Information on likely routes of exposure: - Ingestion: Fatal if swallowed! May cause burns in mouth and throat, burning sensation, cough, laboured breathing, shortness of breath, abdominal pain, vomiting, shock or collapse. Symptoms may be delayed. - Eye contact: Corrosive! Causes serious eye damage, redness, pain, severe burns. - Skin contact: Fatal in contact with skin! Corrosive! Causes severe skin burns, redness, pain, blisters. Hydrofluoric acid penetrates rapidly and deeply below fat layers binding and depleting tissue calcium. - Inhalation: Fatal if inhaled! May cause burns in mouth and throat, burning sensation, cough, laboured breathing, shortness of breath, abdominal pain, vomiting, shock or collapse. Symptoms may be delayed. Chronic effects: Chronic low dose exposure binding and depleting tissue calcium. - Inhalation: Fatal if inhaled! May cause burns in mouth and throat, burning sensation, cough, laboured breathing, shortness of breath, abdominal pain, vomiting, shock or collapse. Symptoms may be delayed. Chronic effects: Chronic low
Acute	· · · · · · · · · · · · · · · · · · ·
Inhalation	Acute toxicity (Inhalation): - LC50, Rat = 996 ppm (1 h) [Supplier's SDS]. - LC50, Mice = 280 mg/m3 (1 h) [NICNAS]. - LC50, Rats = 792 - 1,909 mg/m3 (1 h) [NICNAS].
Carcinogen Category	None

12. ECOLOGICAL INFORMATION

Ecotoxicity	Aquatic toxicity: - EC50, Crustacea (Gammaridea): 73.3 mg/L (96 h) [Calculated from test data of NaF (EC50 = 38.28 mg-F/L/96 hr); Supplier's SDS]. - NOEC, Fish (Atheriniformes): >= 8.6 mg/L (28 d) [Calculated from test data of NaF (NOEC = 9.9 mg/L/28 days); Supplier's SDS].
Persistence/Degradability	Low persistence (water/soil). Low persistence (air).
Mobility	Low mobility in soil (Koc: 14.3).
Environmental Fate	Harmful to aquatic life - Avoid release to the environment.
Bioaccumulation Potential	Low bioaccumulative potential (Log Kow: -1.38).
Environmental Impact	No Data Available

13. DISPOSAL CONSIDERATIONS

General Information	Dispose of contents/container in accordance with local/regional/national regulations.
Special Precautions for Land Fill	Contaminated packaging: Empty containers should be taken to an approved waste handling site for recycling or disposal. Containers may still present chemical hazard when empty.

14. TRANSPORT INFORMATION

Land Transport (Australia) ADG Code	
Proper Shipping Name	HYDROFLUORIC ACID, with not more than 60% hydrogen fluoride
Class	8 Corrosive Substances
Subsidiary Risk(s)	6.1 Toxic and Infectious Substances - Toxic Substances
EPG	40 Toxic And/Or Corrosive Substances Non-Combustible - Water Reactive
UN Number	1790
Hazchem	2X
Pack Group	Ш
Special Provision	No Data Available
Land Transport (Malaysia) ADR Code	
Proper Shipping Name	HYDROFLUORIC ACID, with not more than 60% hydrogen fluoride
Class	8 Corrosive Substances
Subsidiary Risk(s)	6.1 Toxic and Infectious Substances - Toxic Substances
EPG	40 Toxic And/Or Corrosive Substances Non-Combustible - Water Reactive
UN Number	1790
Hazchem	2X
Pack Group	II
Special Provision	No Data Available

Land Transport (New Zealand) NZS5433

Proper Shipping Name	HYDROFLUORIC ACID, with not more than 60% hydrogen fluoride
Class	8 Corrosive Substances
Subsidiary Risk(s)	6.1 Toxic and Infectious Substances - Toxic Substances
EPG	40 Toxic And/Or Corrosive Substances Non-Combustible - Water Reactive
UN Number	1790
Hazchem	2X
Pack Group	II
Special Provision	No Data Available

Land Transport (United States of America) US DOT

Proper Shipping Name	HYDROFLUORIC ACID, with not more than 60% hydrogen fluoride
Class	8 Corrosive Substances
Subsidiary Risk(s)	6.1 Toxic and Infectious Substances - Toxic Substances
ERG	157 Substances - Toxic and/or Corrosive (Non-Combustible / Water-Sensitive)
UN Number	1790
Hazchem	2X
Pack Group	I
Special Provision	No Data Available
Sea Transport IMDG Code	
Proper Shipping Name	HYDROFLUORIC ACID solution, with not more than 60% hydrogen fluoride
Class	8 Corrosive Substances
Subsidiary Risk(s)	6.1 Toxic and Infectious Substances - Toxic Substances
UN Number	1790
Hazchem	2X
Pack Group	II
Special Provision	No Data Available
EMS	F-A, S-B
Marine Pollutant	Νο
Air Transport IATA DGR	
Proper Shipping Name	Hydrofluoric acid 60% or less hydrogen fluoride
Class	8 Corrosive Substances
Subsidiary Risk(s)	6.1 Toxic and Infectious Substances - Toxic Substances
UN Number	1790
Hazchem	2X
Pack Group	II

No Data Available

National Transport Commission (Australia)

Special Provision

Australian Code for the Transport of Dangerous Goods by Road & Rail (ADG Code)

Dangerous Goods Classification

Dangerous Goods according to the criteria of the Australian Code for the Transport of Dangerous Goods by Road & Rail (ADG Code)

15. REGULATORY INFORMATION

General Information Poisons Schedule (Aust)	HYDROFLUORIC ACID Schedule 7	
Environmental Protection Authority (New Zealand) Hazardous Substances and New Organisms Amendment Act 2015		
Approval Code	HSR001568	
National/Regional Inventories		
Australia (AIIC)	Listed	
Canada (DSL)	Not Determined	
Canada (NDSL)	Not Determined	
China (IECSC)	Not Determined	
Europe (EINECS)	Not Determined	
Europe (REACh)	Not Determined	
Japan (ENCS/METI)	Not Determined	
Korea (KECI)	Not Determined	
Malaysia (EHS Register)	Not Determined	
New Zealand (NZIoC)	Listed	
Philippines (PICCS)	Not Determined	
Switzerland (Giftliste 1)	Not Determined	
Switzerland (Inventory of Notified Substances)	Not Determined	
Taiwan (NCSR)	Not Determined	
USA (TSCA)	Not Determined	

16. OTHER INFORMATION

Related Product CodesHYFLAC1013, HYFLAC1015, HYFLAC1016, HYFLAC1035, HYFLAC1057, HYFLAC1100, HYFLAC1101, HYFLAC1200,
HYFLAC1400, HYFLAC1700, HYFLAC1800, HYFLAC2200, HYFLAC2210, HYFLAC2249, HYFLAC3500, HYFLAC3501,
HYFLAC3520, HYFLAC4200, HYFLAC4203, HYFLAC4204, HYFLAC4205, HYFLAC4210, HYFLAC4249, HYFLAC4300,
HYFLAC4400, HYFLAC4401, HYFLAC4449, HYFLAC4450, HYFLAC4500, HYFLAC4600, HYFLAC5000, HYFLAC5001,
HYFLAC5513, HYFLAC8000, HYFLAC9100

Revision	4
Revision Date	23 May 2023
Reason for Issue	Updated SDS
Key/Legend	< Less Than
ney, zegena	> Greater Than
	AICS Australian Inventory of Chemical Substances
	atm Atmosphere
	CAS Chemical Abstracts Service (Registry Number)
	cm ² Square Centimetres
	CO2 Carbon Dioxide
	COD Chemical Oxygen Demand
	deg C (°C) Degrees Celcius
	EPA (New Zealand) Environmental Protection Authority of New Zealand deg F (°F) Degrees Farenheit
	g Grams
	g/cm ³ Grams per Cubic Centimetre
	g/I Grams per Litre
	HSNO Hazardous Substance and New Organism
	IDLH Immediately Dangerous to Life and Health
	immiscible Liquids are insoluable in each other.
	inHg Inch of Mercury
	inH2O Inch of Water
	K Kelvin
	kg Kilogram kg/m³ Kilograms per Cubic Metre
	lb Pound
	LC50 LC stands for lethal concentration. LC50 is the concentration of a material in air which causes the death of 50%
	(one half) of a group of test animals. The material is inhaled over a set period of time, usually 1 or 4 hours.
	LD50 LD stands for Lethal Dose. LD50 is the amount of a material, given all at once, which causes the death of 50% (one
	half) of a group of test animals.
	Itr or L Litre
	m ³ Cubic Metre
	mbar Millibar me Millibram
	mg Milligram mg/24H Milligrams per 24 Hours
	mg/kg Milligrams per Kilogram
	mg/m ³ Milligrams per Cubic Metre
	Misc or Miscible Liquids form one homogeneous liquid phase regardless of the amount of either component present.
	mm Millimetre
	mmH20 Millimetres of Water
	mPa.s Millipascals per Second
	N/A Not Applicable
	NIOSH National Institute for Occupational Safety and Health NOHSC National Occupational Heath and Safety Commission
	OECD Organisation for Economic Co-operation and Development
	Oz Ounce
	PEL Permissible Exposure Limit
	Pa Pascal
	ppb Parts per Billion
	ppm Parts per Million
	ppm/2h Parts per Million per 2 Hours ppm/6h Parts per Million per 6 Hours
	psi Pounds per Square Inch
	R Rankine
	RCP Reciprocal Calculation Procedure
	STEL Short Term Exposure Limit
	TLV Threshold Limit Value
	tne Tonne
	TWA Time Weighted Average
	ug/24H Micrograms per 24 Hours
	UN United Nations wt Weight
	WE WEIGHT