

1. IDENTIFICATION

Product Name Ferric Ammonium Citrate

 Other Names
 Ammonium ferric citrate; Ammonium iron(III) citrate; Citric acid, ammonium iron salt, green

 Uses
 Food additive, Acidity regulator; Water purification; Reducing agent; Medical imaging.

Chemical Family No Data Available
Chemical Formula C6H8O7.xFe.xH3N

Chemical Name 1,2,3-Propanetricarboxylic acid, 2-hydroxy-, ammonium iron(3+) salt

Product Description No Data Available

Contact Details of the Supplier of this Safety Data Sheet

Organisation	Location	Telephone
Redox Ltd	2 Swettenham Road Minto NSW 2566 Australia	+61-2-97333000
Redox Ltd	11 Mayo Road Wiri Auckland 2104 New Zealand	+64-9-2506222
Redox Inc.	3960 Paramount Boulevard Suite 107 Lakewood CA 90712 USA	+1-424-675-3200
Redox Chemicals Sdn Bhd	Level 2, No. 8, Jalan Sapir 33/7	+60-3-5614-2111

Seksyen 33, Shah Alam Premier Industrial Park

Emergency Contact Details

For emergencies only; DO NOT contact these companies for general product advice.

40400 Shah Alam Sengalor, Malaysia

Organisation	Location	Telephone
Poisons Information Centre	Westmead NSW	1800-251525 131126
Chemcall	Australia	1800-127406 +64-4-9179888
Chemcall	Malaysia	+64-4-9179888
Chemcall	New Zealand	0800-243622 +64-4-9179888
National Poisons Centre	New Zealand	0800-764766
CHEMTREC	USA & Canada	1-800-424-9300 CN723420 +1-703-527-3887

2. HAZARD IDENTIFICATION

Poisons Schedule (Aust) Not Scheduled





Globally Harmonised System

Hazard Classification NOT hazardous according to the criteria of the Globally Harmonised System of Classification and Labelling of

Chemicals (GHS)

Signal Word None

National Transport Commission (Australia)

Australian Code for the Transport of Dangerous Goods by Road & Rail (ADG Code)

Dangerous Goods ClassificationNOT Dangerous Goods according to the criteria of the Australian Code for the Transport of Dangerous Goods

by Road & Rail (ADG Code)

Environmental Protection Authority (New Zealand)

Hazardous Substances and New Organisms Amendment Act 2015

HSNO Classifications Health Hazards **6.4A** Substances that are irritating to the eye

3. COMPOSITION/INFORMATION ON INGREDIENTS

Ingredients

Chemical Entity	Formula	CAS Number	Proportion
Ferric ammonium citrate	C6H8O7.xFe.xH3N	1185-57-5	<=100 %

4. FIRST AID MEASURES

Description of necessary measures according to routes of exposure

Swallowed IF SWALLOWED: Rinse mouth, then drink water (two glasses at most). Loosen tight clothing, such as collar, tie, belt or

waist-band. Do NOT induce vomiting unless directed to do so by medical personnel. Get immediate medical

advice/attention if large quantities of this material are swallowed or if you feel unwell. Never give anything by mouth to

an unconscious person.

Eye IF IN EYES: Immediately flush eyes with running water for several minutes, holding eyelids open and occasionally lifting

the upper and lower lids. Remove contact lenses if present and easy to do. Continue rinsing for at least 15 minutes. If eye

irritation persists, get medical advice/attention.

Skin IF ON SKIN: Wash with plenty of soap and water. Take off contaminated clothing and wash it before reuse. If skin irritation

occurs, get medical advice/attention.

Inhaled IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing. Loosen tight clothing,

such as collar, tie, belt or waist-band. If respiratory symptoms persist, get medical advice/attention. Give artificial

respiration if victim is not breathing. Administer oxygen if breathing is difficult.

Advice to Doctor Treat symptomatically.

 $\label{thm:medical conditions Aggravated by} \ \ \mbox{No information available}.$

Exposure

5. FIRE FIGHTING MEASURES

General Measures If safe to do so, move undamaged containers from fire area. Cool containers with water spray until well after fire is out.

Flammability Conditions May be combustible at high temperatures.

Extinguishing Media Use dry chemical, Carbon dioxide (CO2), foam or water spray for extinction - Do not use water jets.

*Use extinguishing measures that are appropriate to local circumstances and the surrounding environment.

Fire and Explosion Hazard Avoid generating dust; fine dust dispersed in air in sufficient concentrations, and in the presence of an ignition source is a

potential dust explosion hazard. Containers may explode when heated. Ambient fire may liberate hazardous vapours.

Development of hazardous combustion gases or vapours possible in the event of fire, including Carbon oxides, Nitrogen

Hazardous Products of Combustion

oxides (NOx), Iron oxides, Ammonia.

Special Fire Fighting Instructions

Contain runoff from fire control or dilution water - Runoff may pollute waterways.

Personal Protective Equipment

Wear positive pressure self-contained breathing apparatus (SCBA). Structural firefighters' protective clothing will only

provide limited protection.

Flash Point
No Data Available
Lower Explosion Limit
No Data Available
Upper Explosion Limit
No Data Available
Auto Ignition Temperature
No Data Available
Hazchem Code
No Data Available

6. ACCIDENTAL RELEASE MEASURES

General Response Procedure Ensure adequate ventilation. ELIMINATE all ignition sources (if dust clouds can occur). Do not touch or walk through

spilled material. Avoid generating dust. Avoid breathing dust and contact with eyes, skin and clothing.

Clean Up Procedures With clean shovel, place material into clean, dry containers and cover loosely; move containers from spill area (see

SECTION 13).

Containment Stop leak if safe to do so. Prevent dust cloud. Cover powder spill with plastic sheet or tarp to minimise spreading.

Decontamination Finish cleaning by spreading water on the contaminated surface and dispose of according to local authority requirements.

Environmental Precautionary

Measures

 $\label{eq:prevent} \mbox{Prevent entry into drains and waterways.}$

Evacuation Criteria Spill or leak area should be isolated immediately. Keep unauthorised personnel away.

Personal Precautionary Measures

Use personal protective equipment as required (see SECTION 8).

*In case of a large spill, a self-contained breathing apparatus should be used to avoid inhalation of the product.

7. HANDLING AND STORAGE

Handling Safety showers and eyewash facilities should be provided within the immediate work area for emergency use. Ensure

adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Avoid formation of dust or aerosols. Avoid breathing dust or aerosols and contact with eyes, skin and clothing. Do not ingest. Use personal protective equipment as required (see SECTION 8). Dry powders can build static electricity charges when subjected to the friction of transfer and mixing operations. Provide adequate precautions, such as electrical grounding and bonding, or

inert atmospheres.

Storage Storage Store in a cool, dry and well-ventilated place, out of direct sunlight. Keep container tightly closed. Protect from moisture

(hygroscopic). Keep away from heat and sources of ignition - No smoking. Keep away from incompatible materials (see

SECTION 10).

Container Store in light-resistant containers.

8. EXPOSURE CONTROLS / PERSONAL PROTECTION

General No specific exposure standards are available for this product. For Iron salts, soluble (as Fe):

Safe Work Australia Exposure Standard: TWA = 1 mg/m3.
 New Zealand Workplace Exposure Standard: TWA = 1 mg/m3.

Exposure Limits No Data Available

Biological Limits No information available.

Engineering Measures A system of local and/or general exhaust is recommended to keep employee exposures as low as possible. Local exhaust

ventilation is generally preferred because it can control the emissions of the contaminant at its source, preventing

dispersion of it into the general work area.

Personal Protection Equipment - Respiratory protection: Not required; except in case of dust or aerosol formation. Recommended: Dust/particulate

respirator (refer to AS/NZS 1715 & 1716).

- Eye/face protection: Wear appropriate eye protection to avoid eye contact. Recommended: Safety glasses or splash

goggles.

- Hand protection: Handle with gloves. Recommended: Protective gloves.

- Skin/body protection: Wear appropriate personal protective clothing to avoid skin contact. Recommended: Lab coat; Full

suit, Boots.

Special Hazards Precaustions No information available.

Work Hygienic Practices Do not eat, drink or smoke when using this product. Wash hands thoroughly after handling. Take off contaminated

clothing and wash it before storage or reuse.

9. PHYSICAL AND CHEMICAL PROPERTIES

Physical StateSolidAppearancePowderOdourOdourless

ColourBrownish, yellowishpHNo Data AvailableVapour PressureNo Data AvailableRelative Vapour DensityNo Data AvailableBoiling PointNo Data AvailableMelting PointNo Data AvailableFreezing PointNo Data Available

Solubility Soluble in water - Practically insoluble in alcohol

Specific Gravity 1.8 (Water = 1)**Flash Point** No Data Available **Auto Ignition Temp** No Data Available **Evaporation Rate** No Data Available **Bulk Density** No Data Available **Corrosion Rate** No Data Available **Decomposition Temperature** No Data Available Density No Data Available Specific Heat No Data Available **Molecular Weight** No Data Available **Net Propellant Weight** No Data Available **Octanol Water Coefficient** No Data Available **Particle Size** No Data Available **Partition Coefficient** No Data Available **Saturated Vapour Concentration** No Data Available Vapour Temperature No Data Available Viscosity No Data Available

Volatile Percent No Data Available
VOC Volume No Data Available

Additional Characteristics No information available.

Potential for Dust Explosion Avoid generating dust; fine dust dispersed in air in sufficient concentrations, and in the presence of an ignition source is a

potential dust explosion hazard.

Fast or Intensely Burning

Characteristics

No information available.

Flame Propagation or Burning Rate of Solid Materials No information available.

Non-Flammables That Could Contribute Unusual Hazards to a No information available.

Properties That May Initiate or Contribute to Fire Intensity

May be combustible at high temperatures.

Reactions That Release Gases or

Development of hazardous combustion gases or vapours possible in the event of fire, including Carbon oxides, Nitrogen

oxides (NOx), Iron oxides, Ammonia.

Release of Invisible Flammable

Vapours and Gases

Vapours

No information available.

10. STABILITY AND REACTIVITY

General Information No information available.

Chemical Stability Stable under recommended storage conditions.

Conditions to Avoid Avoid exposure to light. Keep away from heat and sources of ignition.

Materials to Avoid Incompatible/reactive with oxidising agents, iodides, Acacia preparations and tannins.

Hazardous Decomposition

Products

Development of hazardous combustion gases or vapours possible in the event of fire, including Carbon oxides, Nitrogen

oxides (NOx), Iron oxides, Ammonia.

Hazardous Polymerisation Will not occur.

11. TOXICOLOGICAL INFORMATION

General Information - Acute toxicity: Ingestion can produce gastrointestinal tract irritation with hypomotility, diarrhoea.

- Skin corrosion/irritation: Causes mild skin irritation. Skin contact may cause irritation or rash, particularly with moist skin. Prolonged skin contact may cause irritation. No skin irritation (Rabbit, 4h) [ECHA].

- Eye damage/irritation: May cause eye irritation. Eye contact may cause irritation with redness, tearing and abrasion. No eye irritation (Rabbit) [ECHA].

- Respiratory/skin sensitisation: No information available.

- Germ cell mutagenicity: Negative, Ames test [OECD TG 471 (ECHA)]. Negative, Chromosome aberration test in vitro [OECD TG 473 (ECHA)].

Carcinogenicity: No information available.
Reproductive toxicity: No information available.

- STOT (single exposure): Inhalation of high concentrations of dust may cause nasal, throat or lung irritation. Symptoms may include coughing and wheezing.

- STOT (repeated exposure): Prolonged eye contact may cause a brownish discolouration of the eyes.

- Aspiration toxicity: No information available.

Acute

Ingestion Acute toxicity (Oral):

- LD50, Rat: >2,000 mg/kg

Carcinogen Category None

12. ECOLOGICAL INFORMATION

Ecotoxicity Likely to be not toxic to aquatic microorganisms at environmentally relevant concentrations [ECHA].

Persistence/Degradability Readily biodegradable in water [ECHA].

Mobility No information available.

Environmental Fate Prevent entry into drains and waterways.

Bioaccumulation Potential Low potential for bioaccumulation based on log Kow = 3 [ECHA].

Environmental Impact No Data Available

13. DISPOSAL CONSIDERATIONS

General Information Waste material must be disposed of in accordance with local/regional/national regulation.

Special Precautions for Land Fill Leave chemicals in original containers. No mixing with other waste. Handle uncleaned containers like the product itself.

14. TRANSPORT INFORMATION

Land Transport (Australia)

ADG Code

Proper Shipping Name Ferric Ammonium Citrate

Class No Data Available
Subsidiary Risk(s) No Data Available

No Data Available

UN NumberNo Data AvailableHazchemNo Data AvailablePack GroupNo Data AvailableSpecial ProvisionNo Data Available

Comments NON-DANGEROUS GOODS: Not regulated for LAND transport.

Land Transport (Malaysia)

ADR Code

Proper Shipping Name Ferric Ammonium Citrate

Class No Data Available
Subsidiary Risk(s) No Data Available

No Data Available

UN NumberNo Data AvailableHazchemNo Data AvailablePack GroupNo Data AvailableSpecial ProvisionNo Data Available

Comments NON-DANGEROUS GOODS: Not regulated for LAND transport.

Land Transport (New Zealand)

NZS5433

Proper Shipping Name Ferric Ammonium Citrate

Class No Data Available
Subsidiary Risk(s) No Data Available

No Data Available

UN NumberNo Data AvailableHazchemNo Data AvailablePack GroupNo Data AvailableSpecial ProvisionNo Data Available

Comments NON-DANGEROUS GOODS: Not regulated for LAND transport.

Land Transport (United States of America)

US DOT

Proper Shipping Name Ferric Ammonium Citrate

Class No Data Available
Subsidiary Risk(s) No Data Available

No Data Available

UN NumberNo Data AvailableHazchemNo Data AvailablePack GroupNo Data AvailableSpecial ProvisionNo Data Available

Comments NON-DANGEROUS GOODS: Not regulated for LAND transport.

Sea Transport

IMDG Code

Proper Shipping Name Ferric Ammonium Citrate

Class No Data Available
Subsidiary Risk(s) No Data Available
UN Number No Data Available
Hazchem No Data Available
Pack Group No Data Available
Special Provision No Data Available
EMS No Data Available

Marine Pollutant No

Comments NON-DANGEROUS GOODS: Not regulated for SEA transport.

Air Transport

IATA DGR

Proper Shipping Name Ferric Ammonium Citrate

Class No Data Available
Subsidiary Risk(s) No Data Available
UN Number No Data Available
Hazchem No Data Available
Pack Group No Data Available
Special Provision No Data Available

Comments NON-DANGEROUS GOODS: Not regulated for AIR transport.

National Transport Commission (Australia)

Australian Code for the Transport of Dangerous Goods by Road & Rail (ADG Code)

Dangerous Goods ClassificationNOT Dangerous Goods according to the Criteria of the Australian Code for the Transport of Dangerous Goods

by Road & Rail (ADG Code)

15. REGULATORY INFORMATION

General Information No Data Available
Poisons Schedule (Aust) Not Scheduled

Environmental Protection Authority (New Zealand)

Hazardous Substances and New Organisms Amendment Act 2015

Approval Code HSR002503

HSR004018 (Revoked)

National/Regional Inventories

Australia (AIIC) Listed

Canada (DSL) Not Determined

Canada (NDSL) Not Determined

China (IECSC) Not Determined

Europe (EINECS) 214-686-6

Europe (REACh) Not Determined

Japan (ENCS/METI) Not Determined

Korea (KECI) Not Determined

Malaysia (EHS Register) Not Determined

New Zealand (NZIoC) Listed

Philippines (PICCS) Not Determined

Switzerland (Giftliste 1) Not Determined

Switzerland (Inventory of Notified

Substances)

Not Determined

Taiwan (NCSR) Not Determined

USA (TSCA) Not Determined

16. OTHER INFORMATION

Related Product Codes FEAMCI1000, FEAMCI1001, FEAMCI1002, FEAMCI1200, FEAMCI1200, FEAMCI2000, FEAMCI2005,

FEAMCI2500, FEAMCI2501, FEAMCI3000, FEAMCI3001, FEAMCI3500, FEAMCI4000

Revision

AICS Australian Inventory of Chemical Substances

atm Atmosphere

CAS Chemical Abstracts Service (Registry Number)

cm² Square CentimetresCO2 Carbon Dioxide

COD Chemical Oxygen Demand **deg C (°C)** Degrees Celcius

EPA (New Zealand) Environmental Protection Authority of New Zealand

deg F (°F) Degrees Farenheit

g Grams

g/cm³ Grams per Cubic Centimetre

g/I Grams per Litre

HSNO Hazardous Substance and New Organism **IDLH** Immediately Dangerous to Life and Health **immiscible** Liquids are insoluable in each other.

inHg Inch of Mercury
inH2O Inch of Water

K Kelvin **kg** Kilogram

kg/m3 Kilograms per Cubic Metre

lb Pound

LC50 LC stands for lethal concentration. LC50 is the concentration of a material in air which causes the death of 50% (one half) of a group of test animals. The material is inhaled over a set period of time, usually 1 or 4 hours.

LD50 LD stands for Lethal Dose. LD50 is the amount of a material, given all at once, which causes the death of 50% (one half) of a group of test animals.

Itr or L Litre m³ Cubic Metre mbar Millibar mg Milligram

mg/24H Milligrams per 24 Hours mg/kg Milligrams per Kilogram mg/m³ Milligrams per Cubic Metre

Misc or Miscible Liquids form one homogeneous liquid phase regardless of the amount of either component present.

mm Millimetre

mmH20 Millimetres of Water mPa.s Millipascals per Second

N/A Not Applicable

NIOSH National Institute for Occupational Safety and Health **NOHSC** National Occupational Heath and Safety Commission

OECD Organisation for Economic Co-operation and Development

Oz Ounce

PEL Permissible Exposure Limit

Pa Pascal

ppb Parts per Billion
ppm Parts per Million

ppm/2h Parts per Million per 2 Hours
ppm/6h Parts per Million per 6 Hours

psi Pounds per Square Inch

R Rankine

RCP Reciprocal Calculation Procedure

STEL Short Term Exposure Limit

TLV Threshold Limit Value

tne Tonne

TWA Time Weighted Average **ug/24H** Micrograms per 24 Hours

UN United Nations **wt** Weight

SAFETY DATA SHEET FERRIC AMMONIUM CITRATE REVISION 4, DATE 04 OCT 21